

- L90 ANSWER 21 OF 34 HCAPLUS COPYRIGHT ACS on STN
- AN 2003:183287 HCAPLUS
- ED Entered STN: 11 Mar 2003
- TI Surface functionalization of Si nanoclusters with alkoxides and NMR studies
- AU Zou, Jing; Baldwin, Richard K.; Kauzlarich, Susan M.
- CS Department of Chemistry, University of California, Davis, Davis, CA, 95616, USA
- SO Abstracts of Papers, 225th ACS National Meeting, New Orleans, LA, United States, March 23-27, 2003 (2003), INOR-243 Publisher: American Chemical Society, Washington, D. C. CODEN: 69DSA4
- DT Conference; Meeting Abstract
- LA English
- AB A room temperature reduction route was employed to produce a chloride-capped silicon nanoclusters. These active nanoclusters were readily terminated with various alcs. to give alkoxy-capped silicon nanoclusters, Si-OR (R- OH here=methanol, n-octanol, 1,3-propanediol and benzyl alc.). The terminated products are stable both in organic solvents and isolated as oils. Surface interactions between silicon and the termination groups were characterized by 1H, 13C NMR and FT-IR spectroscopy. The photoluminescence has been used to investigate the various terminated silicon nanoparticles, showing intensive emission peak in the blue region of the spectra. The long-term stability as a function of termination was investigated. The synthesis and characterization will be presented and discussed.



- L118 ANSWER 17 OF 26 HCAPLUS COPYRIGHT ACS on STN
- AN 2000:329679 HCAPLUS
- ED Entered STN: 19 May 2000
- TI Surface-chemical control of optical quenching processes at porous silicon interfaces: Generation of a stable-selective sulfur-dioxide sensor.
- AU Bocarsly, Andrew B.; Wimbish, J. Clint
- CS Department of Chemistry, Princeton University, Princeton, NJ, 08544, USA
- SO Book of Abstracts, 219th ACS National Meeting, San Francisco, CA, March 26-30, 2000 (2000), COLL-414 Publisher: American Chemical Society, Washington, D. C.
  - CODEN: 69CLAC
- DT Conference; Meeting Abstract
- LA English
- Visible photoluminescence from nanoscopic particles of silicon formed by anisotropic etching of single crystal silicon is a well-established phenomenon, A consensus appears to be forming that this process is associated with quantum confined states associated with the Si nanoparticles. Along with this conclusion, a variety of quenching mechanisms have been reported. Previously we indicated that static quenching via dangling bond surface states could be selected by the synthesis of a low quality ultrathin oxide layer on the porous silicon interface. We also demonstrated that such interfaces were selectively quenched by SO2(g). However, the tendency of the oxide interface to further develop in the presence of humid air made the observed quenching process unstable. We now report that, once formed, the Si/SiOx interface can be stabilized even in the presence of water at elevated temps. by modification with a silylfluorocarbon. This interface is permeable to sulfur dioxide allowing for continued quenching by this species over an extended time period, and making possible a functional sulfur dioxide sensor.